

STUDIES ON CATALYTIC HYDROGENATION OF
16-FURFURYLIDENE-17 β -HYDROXY STEROIDS.

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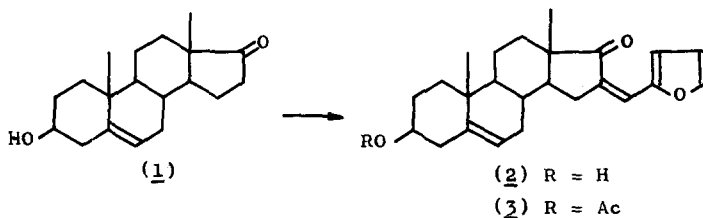
Condensation of androstenolone (1) with furfural in the presence of sodium hydroxide under nitrogen, afforded in quantitative yield 16-furfurylidene-androstenolone (2) - m.p. 190°; $[\alpha]_D^{20} = -107^\circ$ (c=0,57 in CHCl₃); UV-Spectrum: $\lambda_{\max} = 327$ nm ($\epsilon = 26,000$); IR-Spectrum: $\nu_{\max} = 3509, 1724, 1640, 1563, 1026, 1010, 892, 752 \text{ cm}^{-1}$.

Acetylation of (2) with Ac₂O/Py gave the corresponding 3 β -acetate (3) - m.p. 196°; $[\alpha]_D^{20} = -165^\circ$ (c=0,58 in CHCl₃); UV-Spectrum: $\lambda_{\max} = 325$ nm ($\epsilon = 27,600$); IR-Spectrum: $\nu_{\max} = 1724, 1626, 1550, 1250, 1026, 1010, 885, 746 \text{ cm}^{-1}$; NMR-Spectrum**:
 $\delta = 7,60/d$ (J=1,8)/(1); ca.7,2/m/(1); 6,64/d (J=6,3)/(1);

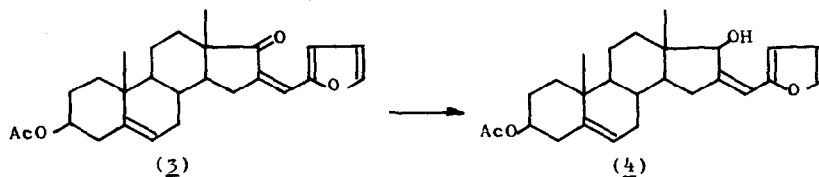
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** Chemical shifts are given in ppm and coupling constants in cps; symbols d, t, q and m indicate doublet, triplet, quartet and multiplet, respectively; the numbers in parentheses represent the number of protons of the corresponding peak.

6,52/q ($J=3,6$ and $1,8$)/(1); 5,44/d ($J=4,0$)/(1); ca.4,6/m (broad)/(1); 2,02/s/(3); 1,09/s/(3); 0,96/s/(3).



Reduction of (2) with sodium borohydride in methanol at 0° , gave a 90% yield of 3β -acetoxy-16-furfurylidene-17 β -hydroxy- Δ^5 -androstene (4) - m.p. 214° ; $[\alpha]_D^{20} = -160^\circ$ ($c=0,64$ in CHCl_3); UV-Spectrum: $\lambda_{\text{max}} = 271 \text{ nm}$ ($\epsilon = 24,700$); IR-Spectrum: $\nu_{\text{max}} = 3425, 1724, 1626$ (shoulder), $1550, 1258, 1026, 1010, 879, 742 \text{ cm}^{-1}$; NMR-Spectrum: $\delta = 7,39/\text{d}$ ($J=1,75$)/(1); ca.6,4/m/(2); ca.6,2/d ($J=3,0$)/(1); 5,40/d ($J=4,0$)/(1); ca.4,6/m (broad)/(1); ca.4,0/s (broad)/(1); 2,02/s/(3); 1,06/s/(3); 0,70/s/(3).



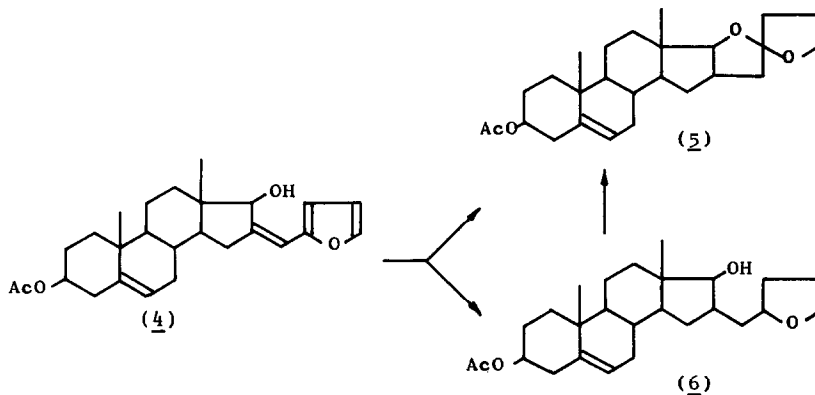
An ethylacetate solution of (4) was hydrogenated under normal pressure using PtO_2 as catalyst. After 6 hours the

absorption of hydrogen ceased and the resulting mixture was chromatographed on a silica-gel column. The following products were isolated and identified:

(a) Spiro-ketal (5) - m.p. 170° ; $[\alpha]_D^{20} = 0^{\circ}$ ($c=1,00$ in CHCl_3); UV-Spectrum: no absorption; IR-Spectrum: $\nu_{\text{max}} = 1722, 1233, 1096, 1080, 1064, 1046, 1017, 990, 975, 920, 903 \text{ cm}^{-1}$; NMR-Spectrum: $\delta = 5,36/\text{d}$ ($J=4,6$)/(1); ca.4,6/m (broad)/(1); 4,05/d ($J=9,8$)/(1); ca.3,9/m/(2); 2,02/s/(3); 1,04/s/(3); 0,74/s/(3).

(b) 3 β -Acetoxy-16 β -tetrahydrofurfuryl-17 β -hydroxy- Δ^5 -androstene (6) - m.p. 210° ; $[\alpha]_D^{20} = -61^{\circ}$ ($c=1,00$ in CHCl_3); UV-Spectrum: no absorption; IR-Spectrum: $\nu_{\text{max}} = 3356, 1720, 1235, 1070 \text{ cm}^{-1}$; NMR-Spectrum: $\delta = 5,40/\text{d}$ ($J=4,0$)/(1); ca.4,6/m (broad)/(1); ca.3,8/m/(4); 3,3/s (broad)/(1); 2,02/s/(3); 1,04/s/(3); 0,78/s/(3).

Oxidation of (6) with lead tetraacetate in benzene gave the spiro-ketal (5).

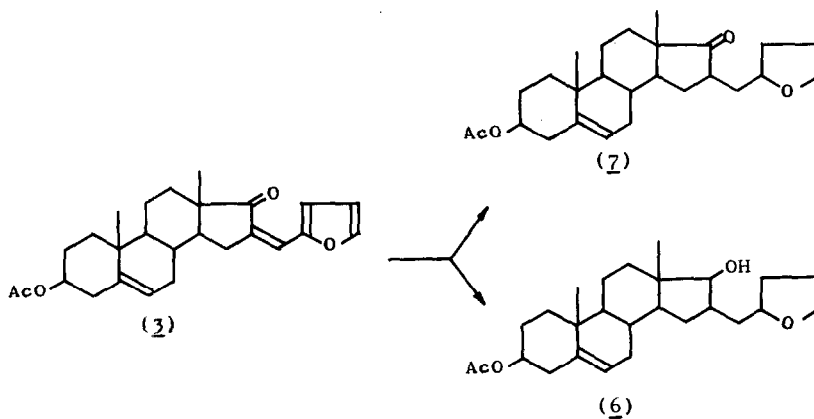


Hydrogenation of 3 β -acetoxy-16-furfurylidene-androst-enolone (3) under the same experimental conditions yielded the following products:

(a) 3 β -Acetoxy-16 β -tetrahydrofurfuryl-17-oxo- Δ^5 -androst-ene (7) - m.p. 176°; $[\alpha]_D^{20} = +25^\circ$ (c=0,60 in CHCl_3); UV-Spectrum: no absorption; IR-Spectrum: $\nu_{\text{max}} = 1736, 1250, 1080 \text{ cm}^{-1}$; NMR-Spectrum: $\delta = 5,40/\text{d}$ (J=4,0)/(1); ca.4,6/m (broad)/(1); ca. 3,8/m/(3); 2,02/s/(3); 1,06/s/(3); 0,80/s/(3).

(b) 3 β -Acetoxy-16 β -tetrahydrofurfuryl-17 β -hydroxy- Δ^5 -androstene (6).

In this case no spiro-ketal was obtained.



Hydrogenation of (3) in ethylacetate under normal pressure with Raney-nickel as catalyst afforded the same products as above, the yield of (7) being higher.

The intramolecular cyclisation of β -furylacrolein derivatives was carried out under a pressure of 100-200 atm. at 160° and with Raney-nickel on Kieselgur as catalyst^{1,2}. In our case, however, the intramolecular cyclisation of (4) into the spiro-ketal (5) occurs by hydrogenation at room temperature with PtO₂ as catalyst. This is probably due both to the rigid conformation of the D-ring and to the cis-configuration of the 17 β -hydroxy and 16 β -dihydrofurfuryl groups (probable intermediate in the reaction), which allows close enough approach of the reacting centers*.

R E F E R E N C E S

1. H.E.Burdick & H.Adkins, J.Am.chem.Soc. 56, 438 (1934)
2. M.Farlow, H.E.Burdick & H.Adkins, J.Am.chem.Soc. 56, 2498 (1934)

* Cyclisation of furfurylidene derivatives of cyclohexanone also requires the conditions proposed by Adkins^{1,2}.